

Unit 4 Covalent bonding - Reading assignment 1 - 16.1 pg 437-451 - Ques. 1-12

K	T	U
ionic compound	single covalent bond	structural formula
molecular compound	electron dot formulas	unshared pairs
diatomic elements	double covalent bond	coordinate covalent bond
polyatomic ions	triple covalent bond	bond dissociation energy
	common covalent compds	resonance structures
		exceptions to the octet rule
		diamagnetic
		paramagnetic

Questions - 16.1 1-12.

1. ~~~~~

2. ~~~~~